

Chemistry Chapter 12 Answer Key

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Answer Key Chapter 12 - Chemistry 2e | OpenStax 1. The instantaneous rate is the rate of a reaction at any particular point in time, a period of time that is so short that the concentrations of reactants and products change by a negligible amount. The initial rate is the instantaneous rate of reaction as it starts (as product just begins to form).

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SHORT ANSWER 1. a. Increasing the pressure of the solute gas above the solution puts stress on the equilibrium of the system. Gas molecules collide with the liquid surface more often, causing an increase in the rate of gas molecules entering into solution. b. As the temperature of the water increases, water molecules move faster,

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